

4-Fluoro-N-[3-(2-fluorophenyl)-4-methyl-2,3-dihydro-2-thienylidene]-benzamide

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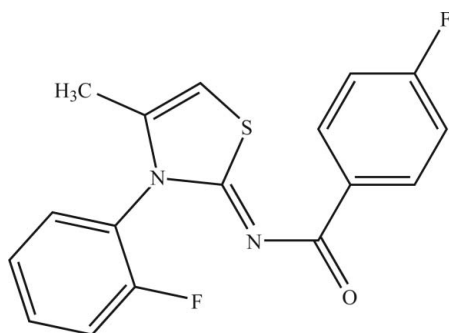
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.034; wR factor = 0.082; data-to-parameter ratio = 16.1.

In the title compound, $\text{C}_{17}\text{H}_{12}\text{F}_2\text{N}_2\text{OS}$, the planar thiazole ring (r.m.s. deviation = 0.012 Å) makes dihedral angles of 15.08 (9) and 81.81 (6)° with the 4-fluorophenyl and 2-fluorophenyl rings, respectively. The 2-fluorophenyl ring is disordered over two orientations with site-occupancy factors of 0.810 (3) and 0.190 (3). The structure contains intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For the biological activity of imino-1,3-thiazoline derivatives, see: Kim *et al.* (2007); Vicini *et al.* (2006); Hosseinimehr *et al.* (2001); Zhang *et al.* (2000); Pietrancosta *et al.* (2006). For details of the synthesis, see: Saeed *et al.* (2008a). For a related structure, see: Saeed *et al.* (2008b).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{12}\text{F}_2\text{N}_2\text{OS}$
 $M_r = 330.35$
 Orthorhombic, $P2_12_12_1$
 $a = 7.0982$ (14) Å
 $b = 11.423$ (2) Å
 $c = 18.949$ (4) Å
 $V = 1536.5$ (5) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.24$ mm⁻¹
 $T = 173$ K
 $0.36 \times 0.34 \times 0.28$ mm

Data collection

Stoe IPDS-II two-circle diffractometer
 Absorption correction: multi-scan (*MULABS*; Spek, 2009; Blessing, 1995)
 $T_{\min} = 0.920$, $T_{\max} = 0.937$
 10484 measured reflections
 3531 independent reflections
 3213 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.082$
 $S = 0.99$
 3531 reflections
 219 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.15$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³
 Absolute structure: Flack (1983), 1491 Friedel pairs
 Flack parameter: -0.15 (6)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C5}-\text{H5}\cdots\text{O1}^i$	0.95	2.41	3.322 (2)	160

Symmetry code: (i) $-x + 2, y + \frac{1}{2}, -z + \frac{3}{2}$.

Data collection: *X-Area* (Stoe & Cie, 2001); cell refinement: *X-Area*; data reduction: *X-Area*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BI2375).

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supplementary materials

Acta Cryst. (2009). E65, o1643 [doi:10.1107/S1600536809022314]

4-Fluoro-N-[3-(2-fluorophenyl)-4-methyl-2,3-dihydro-2-thienylidene]benzamide

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Comment

The imino-1,3-thiazoline group is found in a variety of biologically active natural products and finds extensive applications in medicinal chemistry. 2-Thiazolylimino-5-arylidene-4-thiazolidinones show noticeable antimicrobial activity against bacteria, yeasts and mould (Kim *et al.*, 2007). 3-Substituted 2-(cyanoimino)thiazolidines can be used in agriculture due to their neonicotinoid insecticidal activity (Vicini *et al.*, 2006). 3-Substituted thiazolidines show radioprotective properties against γ -radiation (Hosseimehr *et al.*, 2001). KHG22394, a 2-imino-1,3-thiazoline derivative, significantly inhibits melanin production in a dose-dependent manner, thus acting as a skin whitening agent (Zhang *et al.*, 2000) and pifithrin- α , another iminothiazoline, is a reversible inhibitor of p53-mediated apoptosis and p53-dependent gene transcription (Pietrancosta *et al.*, 2006).

Experimental

The title compound was prepared according to the procedure reported earlier (Saeed *et al.* (2008a)). Crystallization of the residue in CHCl_3 afforded the title compound (81%) as white needles: Anal. calcd. for $\text{C}_{14}\text{H}_{12}\text{ClNO}$: C 68.44, H 4.92, N 5.70%; found: C 68.39, H 4.90, N 5.67%

Refinement

H atoms were geometrically positioned and refined using a riding model with fixed individual displacement parameters [$U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ or $1.5 U_{\text{eq}}(\text{C}_{\text{methyl}})$] using a riding model with $\text{C}-\text{H}(\text{aromatic}) = 0.95 \text{ \AA}$ or $\text{C}-\text{H}(\text{methyl}) = 0.98 \text{ \AA}$. The *ortho*-fluoro-phenyl ring is disordered over two positions with site occupation factors of 0.810 (3) and 0.190 (3).

Figures

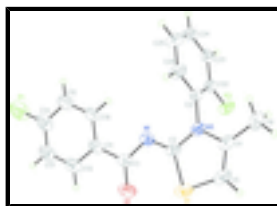


Fig. 1. Perspective view of the title compound with the atom numbering scheme; displacement ellipsoids are at the 50% probability level; H atoms are drawn as small spheres of arbitrary radii. The atoms of the minor occupied sites have been omitted for clarity.

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Crystal data

$\text{C}_{17}\text{H}_{12}\text{F}_2\text{N}_2\text{O}$

$M_r = 330.35$

$F_{000} = 680$

$D_x = 1.428 \text{ Mg m}^{-3}$

supplementary materials

Orthorhombic, $P2_12_12_1$
Hall symbol: P 2ac 2ab
 $a = 7.0982$ (14) Å
 $b = 11.423$ (2) Å
 $c = 18.949$ (4) Å
 $V = 1536.5$ (5) Å³
 $Z = 4$

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 9675 reflections
 $\theta = 3.4$ – 27.8°
 $\mu = 0.24$ mm⁻¹
 $T = 173$ K
Block, colourless
 $0.36 \times 0.34 \times 0.28$ mm

Data collection

Stoe IPDS-II two-circle diffractometer
Radiation source: fine-focus sealed tube
Monochromator: graphite
 $T = 173$ K
 ω scans
Absorption correction: multi-scan (MULABS; Spek, 2009; Blessing, 1995)
 $T_{\min} = 0.920$, $T_{\max} = 0.937$
10484 measured reflections

3531 independent reflections
3213 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$
 $\theta_{\max} = 27.6^\circ$
 $\theta_{\min} = 3.4^\circ$
 $h = -8 \rightarrow 9$
 $k = -12 \rightarrow 14$
 $l = -24 \rightarrow 23$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.082$
 $S = 0.99$
3531 reflections
219 parameters
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.052P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.15$ e Å⁻³
 $\Delta\rho_{\min} = -0.26$ e Å⁻³
Extinction correction: none
Absolute structure: Flack (1983), 1491 Friedel pairs
Flack parameter: -0.15 (6)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.86255 (5)	0.65177 (5)	0.73205 (2)	0.04103 (12)	
N1	0.56925 (19)	0.59033 (13)	0.64274 (7)	0.0316 (3)	
O1	0.7594 (2)	0.44060 (14)	0.68401 (8)	0.0520 (4)	
F1	0.0840 (3)	0.15356 (14)	0.55262 (9)	0.0825 (5)	
F2	0.71228 (15)	0.86728 (14)	0.53579 (6)	0.0406 (4)	0.810 (3)
F2'	0.2818 (7)	0.7551 (6)	0.7021 (3)	0.048 (2)	0.190 (3)
C1	0.6168 (2)	0.47540 (16)	0.65201 (8)	0.0356 (4)	
C2	0.67222 (19)	0.66920 (16)	0.67514 (8)	0.0303 (3)	
N3	0.63342 (18)	0.78545 (13)	0.66789 (7)	0.0304 (3)	
C4	0.7456 (2)	0.86166 (18)	0.70871 (8)	0.0340 (4)	
C5	0.8742 (2)	0.80204 (19)	0.74587 (9)	0.0414 (4)	
H5	0.9626	0.8381	0.7766	0.050*	
C6	0.7117 (3)	0.98996 (19)	0.70565 (10)	0.0436 (4)	
H6A	0.7834	1.0287	0.7432	0.065*	
H6B	0.5771	1.0056	0.7121	0.065*	
H6C	0.7523	1.0201	0.6597	0.065*	
C11	0.4788 (3)	0.39042 (17)	0.62212 (8)	0.0353 (4)	
C12	0.5145 (3)	0.27036 (19)	0.62624 (10)	0.0451 (4)	
H12	0.6301	0.2435	0.6457	0.054*	
C13	0.3827 (3)	0.19005 (18)	0.60216 (10)	0.0513 (5)	
H13	0.4062	0.1083	0.6052	0.062*	
C14	0.2173 (3)	0.2315 (2)	0.57387 (11)	0.0529 (5)	
C15	0.1774 (3)	0.3489 (2)	0.56784 (11)	0.0519 (5)	
H15	0.0622	0.3747	0.5476	0.062*	
C16	0.3098 (3)	0.42878 (17)	0.59203 (10)	0.0405 (4)	
H16	0.2854	0.5103	0.5881	0.049*	
C21	0.48714 (19)	0.82453 (15)	0.62105 (8)	0.0278 (3)	
C22	0.5303 (2)	0.86227 (15)	0.55392 (8)	0.0290 (3)	
H22	0.6587	0.8661	0.5400	0.035*	0.190 (3)
C23	0.3925 (2)	0.89464 (15)	0.50637 (8)	0.0325 (3)	
H23	0.4250	0.9201	0.4602	0.039*	
C24	0.2054 (2)	0.88918 (16)	0.52746 (9)	0.0348 (4)	
H24	0.1085	0.9107	0.4954	0.042*	
C25	0.1592 (2)	0.8527 (2)	0.59472 (9)	0.0439 (4)	
H25	0.0309	0.8499	0.6089	0.053*	
C26	0.2997 (2)	0.8203 (2)	0.64164 (9)	0.0403 (4)	
H26	0.2675	0.7945	0.6877	0.048*	0.810 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.02961 (17)	0.0565 (3)	0.0369 (2)	0.00724 (19)	-0.00870 (15)	0.0061 (2)
N1	0.0334 (6)	0.0324 (8)	0.0290 (6)	0.0029 (5)	-0.0029 (5)	0.0041 (6)
O1	0.0474 (7)	0.0475 (9)	0.0610 (8)	0.0136 (6)	-0.0158 (6)	0.0073 (7)

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F1	0.1076 (11)	0.0365 (7)	0.1034 (11)	-0.0068 (8)	-0.0504 (9)	-0.0073 (8)
F2	0.0270 (6)	0.0573 (10)	0.0374 (6)	-0.0039 (5)	0.0046 (4)	0.0079 (6)
F2'	0.032 (3)	0.074 (5)	0.038 (3)	0.005 (3)	0.006 (2)	0.014 (3)
C1	0.0391 (8)	0.0375 (9)	0.0302 (7)	0.0106 (8)	-0.0003 (6)	0.0046 (7)
C2	0.0263 (6)	0.0403 (10)	0.0242 (6)	0.0023 (6)	-0.0005 (5)	0.0036 (6)
N3	0.0260 (5)	0.0362 (7)	0.0290 (6)	-0.0027 (6)	-0.0054 (5)	0.0023 (5)
C4	0.0269 (6)	0.0454 (11)	0.0299 (7)	-0.0076 (7)	-0.0004 (5)	-0.0033 (7)
C5	0.0288 (7)	0.0603 (12)	0.0351 (8)	-0.0059 (7)	-0.0074 (6)	-0.0013 (7)
C6	0.0402 (8)	0.0462 (12)	0.0443 (9)	-0.0112 (8)	-0.0037 (7)	-0.0079 (8)
C11	0.0460 (9)	0.0325 (9)	0.0275 (7)	0.0087 (7)	-0.0001 (6)	0.0021 (6)
C12	0.0590 (11)	0.0372 (10)	0.0391 (9)	0.0129 (9)	-0.0036 (8)	-0.0004 (8)
C13	0.0751 (13)	0.0302 (10)	0.0485 (10)	0.0111 (9)	-0.0098 (10)	-0.0064 (8)
C14	0.0751 (13)	0.0332 (11)	0.0504 (11)	-0.0010 (10)	-0.0168 (10)	-0.0057 (9)
C15	0.0656 (11)	0.0348 (10)	0.0553 (11)	0.0055 (9)	-0.0237 (9)	-0.0007 (10)
C16	0.0525 (10)	0.0295 (9)	0.0395 (8)	0.0064 (7)	-0.0111 (7)	0.0015 (8)
C21	0.0262 (6)	0.0285 (8)	0.0288 (7)	-0.0004 (6)	-0.0037 (5)	0.0016 (6)
C22	0.0280 (6)	0.0278 (8)	0.0314 (7)	-0.0019 (6)	0.0010 (5)	0.0002 (6)
C23	0.0409 (8)	0.0299 (8)	0.0267 (7)	-0.0024 (7)	-0.0022 (6)	0.0021 (6)
C24	0.0335 (7)	0.0363 (9)	0.0346 (8)	0.0029 (7)	-0.0097 (6)	-0.0014 (7)
C25	0.0258 (6)	0.0679 (13)	0.0379 (8)	0.0020 (8)	-0.0020 (6)	0.0010 (9)
C26	0.0290 (7)	0.0615 (13)	0.0305 (7)	-0.0027 (8)	-0.0009 (6)	0.0070 (8)

Geometric parameters (Å, °)

S1—C5	1.738 (2)	C12—C13	1.388 (3)
S1—C2	1.7400 (15)	C12—H12	0.950
N1—C2	1.313 (2)	C13—C14	1.375 (3)
N1—C1	1.367 (2)	C13—H13	0.950
O1—C1	1.245 (2)	C14—C15	1.376 (3)
F1—C14	1.360 (3)	C15—C16	1.387 (3)
F2—C22	1.3379 (18)	C15—H15	0.950
F2'—C26	1.373 (6)	C16—H16	0.950
C1—C11	1.491 (3)	C21—C22	1.378 (2)
C2—N3	1.363 (2)	C21—C26	1.387 (2)
N3—C4	1.411 (2)	C22—C23	1.380 (2)
N3—C21	1.4370 (18)	C22—H22	0.950
C4—C5	1.339 (2)	C23—C24	1.388 (2)
C4—C6	1.486 (3)	C23—H23	0.950
C5—H5	0.950	C24—C25	1.380 (2)
C6—H6A	0.980	C24—H24	0.950
C6—H6B	0.980	C25—C26	1.386 (2)
C6—H6C	0.980	C25—H25	0.950
C11—C12	1.397 (3)	C26—H26	0.950
C11—C16	1.399 (2)		
C5—S1—C2	90.99 (8)	F1—C14—C13	119.0 (2)
C2—N1—C1	117.49 (14)	F1—C14—C15	118.1 (2)
O1—C1—N1	124.75 (18)	C13—C14—C15	122.9 (2)
O1—C1—C11	120.75 (17)	C14—C15—C16	118.31 (19)
N1—C1—C11	114.48 (14)	C14—C15—H15	120.8

N1—C2—N3	120.60 (13)	C16—C15—H15	120.8
N1—C2—S1	130.07 (14)	C15—C16—C11	120.64 (18)
N3—C2—S1	109.32 (12)	C15—C16—H16	119.7
C2—N3—C4	115.58 (14)	C11—C16—H16	119.7
C2—N3—C21	120.72 (13)	C22—C21—C26	118.93 (14)
C4—N3—C21	123.70 (15)	C22—C21—N3	120.45 (13)
C5—C4—N3	111.06 (17)	C26—C21—N3	120.56 (14)
C5—C4—C6	129.23 (16)	F2—C22—C21	117.72 (14)
N3—C4—C6	119.71 (15)	F2—C22—C23	120.34 (14)
C4—C5—S1	113.00 (13)	C21—C22—C23	121.94 (14)
C4—C5—H5	123.5	C21—C22—H22	118.9
S1—C5—H5	123.5	C23—C22—H22	119.1
C4—C6—H6A	109.5	C22—C23—C24	118.55 (15)
C4—C6—H6B	109.5	C22—C23—H23	120.7
H6A—C6—H6B	109.5	C24—C23—H23	120.7
C4—C6—H6C	109.5	C25—C24—C23	120.43 (14)
H6A—C6—H6C	109.5	C25—C24—H24	119.8
H6B—C6—H6C	109.5	C23—C24—H24	119.8
C12—C11—C16	119.08 (18)	C24—C25—C26	120.12 (14)
C12—C11—C1	119.91 (16)	C24—C25—H25	119.9
C16—C11—C1	120.96 (16)	C26—C25—H25	119.9
C13—C12—C11	120.54 (18)	F2'—C26—C25	127.8 (3)
C13—C12—H12	119.7	F2'—C26—C21	110.0 (3)
C11—C12—H12	119.7	C25—C26—C21	120.02 (15)
C14—C13—C12	118.5 (2)	C25—C26—H26	119.9
C14—C13—H13	120.8	C21—C26—H26	120.0
C12—C13—H13	120.8		
C2—N1—C1—O1	-6.3 (3)	C12—C13—C14—C15	0.6 (3)
C2—N1—C1—C11	171.87 (14)	F1—C14—C15—C16	177.5 (2)
C1—N1—C2—N3	179.61 (14)	C13—C14—C15—C16	-0.6 (4)
C1—N1—C2—S1	-2.0 (2)	C14—C15—C16—C11	-0.4 (3)
C5—S1—C2—N1	-176.30 (15)	C12—C11—C16—C15	1.3 (3)
C5—S1—C2—N3	2.19 (11)	C1—C11—C16—C15	-176.43 (18)
N1—C2—N3—C4	176.05 (13)	C2—N3—C21—C22	-97.29 (18)
S1—C2—N3—C4	-2.61 (15)	C4—N3—C21—C22	82.51 (19)
N1—C2—N3—C21	-4.1 (2)	C2—N3—C21—C26	80.0 (2)
S1—C2—N3—C21	177.21 (10)	C4—N3—C21—C26	-100.2 (2)
C2—N3—C4—C5	1.65 (19)	C26—C21—C22—F2	178.95 (18)
C21—N3—C4—C5	-178.16 (13)	N3—C21—C22—F2	-3.7 (2)
C2—N3—C4—C6	-178.75 (14)	C26—C21—C22—C23	-0.8 (3)
C21—N3—C4—C6	1.4 (2)	N3—C21—C22—C23	176.59 (16)
N3—C4—C5—S1	0.14 (17)	F2—C22—C23—C24	-179.43 (18)
C6—C4—C5—S1	-179.40 (14)	C21—C22—C23—C24	0.3 (3)
C2—S1—C5—C4	-1.36 (13)	C22—C23—C24—C25	0.4 (3)
O1—C1—C11—C12	-3.8 (3)	C23—C24—C25—C26	-0.5 (3)
N1—C1—C11—C12	177.96 (16)	C24—C25—C26—F2'	-161.8 (4)
O1—C1—C11—C16	173.91 (17)	C24—C25—C26—C21	0.0 (3)
N1—C1—C11—C16	-4.3 (2)	C22—C21—C26—F2'	165.4 (3)
C16—C11—C12—C13	-1.3 (3)	N3—C21—C26—F2'	-12.0 (4)

supplementary materials

C1—C11—C12—C13	176.43 (16)	C22—C21—C26—C25	0.6 (3)
C11—C12—C13—C14	0.4 (3)	N3—C21—C26—C25	-176.76 (18)
C12—C13—C14—F1	-177.5 (2)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C5—H5 \cdots O1 ⁱ	0.95	2.41	3.322 (2)	160

Symmetry codes: (i) $-x+2, y+1/2, -z+3/2$.

Fig. 1

